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## Mass, Volume, and Density Notes

**Definitions:**

**Matter:** anything that has a mass and a volume.

**Mass:** the amount of matter in an object. The amount of "stuff" is measured in grams.

**Volume:** the amount of space an object takes up. We use  $\text{cm}^3$  or mL to measure volume.

The formula to calculate volume for a rectangular prism is:

$$L \times W \times H = \text{cm}^3$$

For an irregularly shaped object we use a Graduated Cylinder and water displacement. The volume is found by:

$$\text{Final reading in mL} - \text{the starting reading in mL}$$

**Density:** the ratio of mass to volume. Everything has density; it is how much "stuff" is packed into a given volume. The formula used to find density is:

$$\text{Density} = \text{Mass} \div \text{Volume (D=M/V)}$$

The unit of measurement is  $\text{g/cm}^3$ .

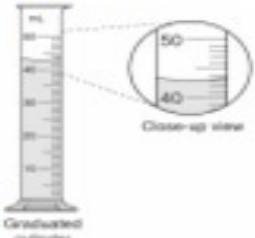

The density of water is  $1.0 \text{ g/cm}^3$ .

**Volume of 1 cubic centimeter (1 cm<sup>3</sup>)**

A cube measuring 1 cm x 1 cm x 1 cm takes up 1 cm<sup>3</sup> of space, holds 1 mL of water and has a mass of 1 gram.

$$1 \text{ mL} = 1 \text{ cm}^3 = 1 \text{ g}$$

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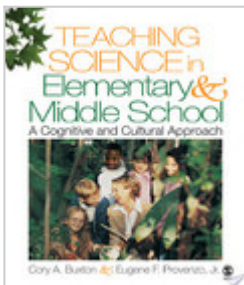
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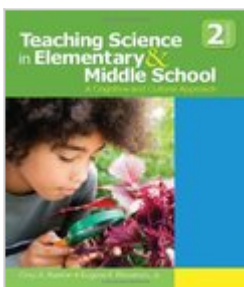
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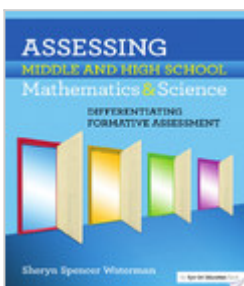
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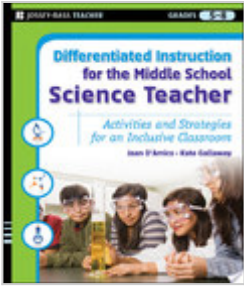
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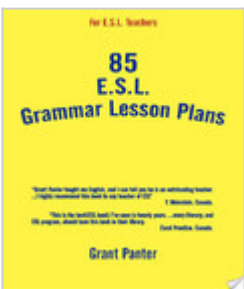
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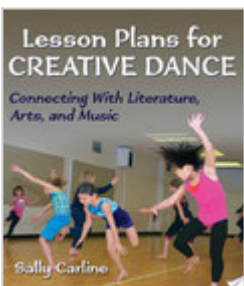
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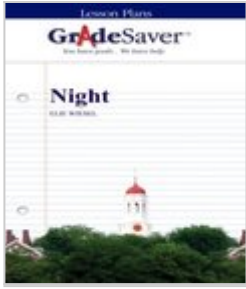
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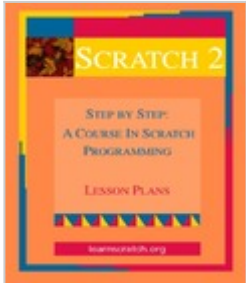
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